

Department of Chemistry & Biochemistry 1068 W. Sheridan Rd. Chicago, IL 60660 <u>https://www.luc.edu/chemistry/</u>

Course:	General Chemistry CHEM 101				
	Semester:Fall 2019Lecture:Section 025 - T/Th 8:30 - 9:45 AM, Cuneo 109Discussions:Section 026 - M 9:20 - 10:10 AM, FH 105Section 027 - M 10:25 - 11:15 AM FH 105Section 028 - M 11:30-12:20 PM Dumbach 230**you must attend your assigned discussion section due to seating limitations				
Professor:	Dr. Caitlin G. Decker, PhD Office: FH 200A Office Hours: T/Th 10-11 AM				
	Email: <u>cdecker@luc.edu</u> ** No specific problem-solving questions will be answered via email. All such questions should be posted to the discussion board (sakaii) so that they are visible to all students or asked during discussion section / office hours.				
Materials:	Textbook Brown, LeMay, et. al. (2018) <u>Chemistry the Central Science</u> , 14 th edition. Print or electronic version is fine. Mastering Chemistry material is optional (there are a lot of resources there). Earlier editions are acceptable. <u>ISBNs:</u> MyLab plus Etext: 9780134553108 MyLab plus loose-leaf: 9780134557328 MyLab plus Hardcopy: 9780134292816				
Sakaii:	All students are enrolled in the class Sakaii site. It is imperative that you check this site daily to keep informed of all activities.				
Important Dates:	Oct 25^{th} – Midterm Grades / Academic Alerts (<i>prior</i> to this date!) Nov 1^{st} – drop deadline				
Exams:	Exam 1 – Tues Sept 24 th Exam 2 – Tues Oct 22 nd Exam 3 – Tues Nov 19 th FINAL - Saturday, Dec 14 th , 9-11 AM, Cuneo 109 *Final Exam IS Cumulative **There will be NO regrades for this course on any exam. Grades are final. You must show your ID to the instructor and sign-in next to your name for each exam. All electronic devices must be turned off and inside bags.				

Course Description:	Lecture and discussion. Basic chemical principles. Topics include atomic and molecular structures, states of matter, energetics and stoichiometry of reactions. For non-chemistry majors and students in the B.A. Chemistry program.					
Prerequisite: Co-requisite:	Math 117 (C- or better) or passing the Loyola math proficiency exam CHEM 111					
Grading Scale:	$\begin{array}{cccc} 93-100\% = A & 90-92\% = A-\\ 87-89\% = B+ & 83-86\% = B & 80-82\% = B-\\ 77-79\% = C+ & 73-76\% = C & 70-72\% = C-\\ 60-69\% = D & & \\ Below \ 60\% = F & \\ **the \ professor \ reserves \ the \ right \ to \ implement \ a \ curve, \ as \ necessary \end{array}$					
Grade:	 Grades will be determined using one of the two methods below (whichever results in a <i>higher</i> overall grade): 1) All three midterms + final are averaged. Thus, each exam will weigh 1/4. 2) The top two mid-term exams weigh 1/4 each, and the final will weigh 1/2. This equates to the final exam score replacing the lowest midterm score. **<i>due to this policy there will be NO make-up exams. If you miss an exam, it will count as the "dropped" exam, and method #2 will be used to calculate the grade.</i> To calculate what you need on the Final: Ex 1) Student X wants to calculate the grade needed on the final exam in order to gain an overall score of 70% or a C- in the class. Student X has received the following scores thus far: Exam 1: 56% Exam 2: 70% Exam 3: 42% Method 1: (56+70+42+N)/4 = 70 multiply by 4 on each side to give: 56+70+42+N)/4=70 Multiply by 4 on each side to give: 56+70+2N)/4=70 Multiply by 4 on each side to give: 56+70+2N)/4=70 Multiply by 4 on each side to give: 56+70+2N)/4=70 Multiply by 4 on each side to give: 56+70+2N)/4=70 Multiply by 4 on each side to give: 56+70+2N)/4=70 Multiply by 4 on each side to give: 56+70+2N)/4=70 Multiply by 4 on each side to give: 56+70+2N)/4=70 Multiply by 4 on each side to give: 56+70+2N=280 Subtract the 2 known scores to give 2N=154 Divide by 2 on each side N=77% Therefore, Student X needs to earn a score of 77% on the final exam in order to 					

Course Content:

- Ch 1. Matter, Energy, and Measurement.
- Ch 2. Atoms, Molecules, and Ions
- Ch 3. Chemical Reactions and Reaction Stoichiometry
- Ch 4. Reactions in Aqueous Solution
- Ch 5. Thermochemistry
- Ch 6. Electronic structure of atoms
- Ch 7. Periodic Properties of the Elements
- Ch 8. Basic Concepts of Chemical Bonding
- Ch 9. Molecular Geometry and Bonding Theories
- Ch 10. Gases
- Ch 24. Organic & Biological Chem. (Selected Topics)

Institutional Policies:

Loyola Official Academic Calendar: www.luc.edu/academics/schedules

Incomplete Grade:

If the Final Exam is missed for extenuating circumstances (incapacitating illness, immediate family member death, fire/flood or related emergency) students must fill-out an "Incomplete Grade Form". Be aware that the option to apply for an incomplete grade is at the discretion of the professor. Incomplete grade info: <u>https://www.luc.edu/regrec/faculty.shtml</u>

Course Repeat Rule:

Effective with the Fall 2017 semester, students are allowed only THREE attempts to pass Chemistry courses with a C- or better grade. The three attempts include withdrawals (W). After the second attempt, the student must secure approval for a third attempt. Students must come to the Chemistry Department, fill out a permission to register form or print it from Depart of Chemistry & Biochemistry website: http://www.luc.edu/chemistry/forms/ and obtain a signature from the Undergraduate Program Director, Assistant Chairperson, or Chairperson in Chemistry. A copy of this form is then taken to your Academic Advisor in Sullivan to secure final permission for the attempt. Students are encouraged to seek help with the course material early and often during the semester. Attend office hours regularly for assistance before any deficiencies become serious!

Accommodation Requests:

Additional time on exams, a quiet space for exams, a note-taker, or permission to record lectures can be requested for qualifying students. It is the responsibility of the student to register with SAC and to provide documentation to the professor prior to the initiation of such accommodations.

Student Accessibility Center: https://www.luc.edu/sac/registerwithsac/

<u>Tentative Course Schedule/Outline:</u> The instructor reserves the right to adjust the schedule and assignments as circumstances may warrant during the semester.

Week	Monday	Tuesday	Wednesday	Thursday	Friday
1	Aug 26 th	Aug 27 th	Aug 28 th	Aug 29 th	Aug 30 th
		Syllabus / Ch. 1		Ch. 1	
2	Sept 2 nd	Sept 3 rd	Sept 4 th	Sept 5 th	Sept 6 th
	LABOR DAY	Ch. 2		Ch. 3	
3	Sept 9 th	Sept 10 th	Sept 11 th	Sept 12 th	Sept 13 th
		Ch. 3		Ch. 4	
4	Sept 16 th	Sept 17 th	Sept 18 th	Sept 19 th	Sept 20 th
		Ch. 4		Review / Catch-up	
5	Sept 23 rd	Sept 24 th	Sept 25 th	Sept 26 th	Sept 27 th
		EXAM 1		Ch. 5	
6	Sept 30 th	Oct 1^{st}	Oct 2^{nd}	Oct 3^{rd}	$Oct \ 4^{th}$
		Ch. 5		Ch. 6	
7	Oct 7 th	Oct 8 th	$Oct 9^{th}$	$Oct \ 10^{th}$	$Oct 11^{th}$
	FALL BREAK			Ch. 6	
8	$Oct \ 14^{th}$	$Oct \ 15^{th}$	$Oct \ 16^{th}$	$Oct \ 17^{th}$	$Oct \ 18^{th}$
		Ch. 7		Review / Catch-up	
9	$Oct 21^{st}$	Oct 22 nd	$Oct \ 23^{rd}$	$Oct 24^{th}$	$Oct \ 25^{th}$
		EXAM 2		Ch. 8	
10	$Oct \ 28^{th}$	Oct 29 th	$Oct \ 30^{th}$	$Oct 31^{st}$	Nov 1 st
	-	Ch. 8		Ch. 9	
11	Nov 4^{th}	Nov 5 th	Nov 6^{th}	Nov 7 th	Nov 8^{th}
		Ch. 9		Ch. 10	· · · ·
12	Nov 11 th	Nov 12 th	Nov 13 th	Nov 14 th	Nov 15 th
		Ch. 10		Review / Catch-up	
13	Nov 18 th	Nov 19 th	Nov 20 th	Nov 21 st	Nov 22 nd
	4	EXAM 3	.7	Ch 24	.1
14	Nov 25 th	Nov 26 th	Nov 27 th	Nov 28 th	Nov 29 th
	7	7	THANKSGIVING BREAK		
15	Dec 2^{nd}	Dec 3 rd	Dec 4^{th}	Dec 5^{th}	Dec 6^{th}
		Ch 24		Review / Catch-up	
16	Dec 9 th	Dec 10^{th}	Dec 11 th	Dec 12 th	Dec 13 th
	Final Exam Week				

FINAL EXAM <mark>Saturday Dec 14th (1996) (19977) (19977) (19977) (19977) (19977) (19977) (19977) (1</mark> <mark>9-11 AM</mark>